

Separation of the centre of mass coordinates for an N-electron atom.

Razmotrimo atom (jon) koji sadrži nucleus mase M , naelektrisanje Ze , i N elektrona mase m .

Obelježimo sa \vec{R}_0 radius vektor nukleusa u odnosu na fiksni koord. početak O , i $\vec{R}_1, \vec{R}_2, \dots, \vec{R}_N$ za elektrone. Hamiltonijan za ovaj sistem je

$$H = T + V$$

$$T = -\frac{\hbar^2}{2M} \nabla_{R_0}^2 + \sum_{i=1}^N \left(-\frac{\hbar^2}{2m} \nabla_{R_i}^2 \right)$$

V - pot. energija kao suma Kulonovih interakcija između $(N+1)$ čestice sistema

Da bi odvojili kretanje centra mase uvedimo skupinu $(\vec{R}_0, \vec{R}_1, \dots, \vec{R}_N) \rightarrow (\vec{R}, \vec{r}_1, \dots, \vec{r}_N)$

$$\vec{R} = \frac{1}{M + Nm} (M\vec{R}_0 + m\vec{R}_1 + \dots + m\vec{R}_N)$$

$$\vec{r}_i = \vec{R}_i - \vec{R}_0, \quad i = 1, 2, \dots, N$$

$$\psi(\vec{R}_0, \vec{R}_1, \dots, \vec{R}_N) \rightarrow \psi(\vec{R}, \vec{r}_1, \dots, \vec{r}_N)$$

$$\frac{\partial \psi}{\partial \vec{R}_0} = \frac{\partial \psi}{\partial \vec{R}} \frac{\partial \vec{R}}{\partial \vec{R}_0} + \sum_{i=1}^N \frac{\partial \psi}{\partial \vec{r}_i} \frac{\partial \vec{r}_i}{\partial \vec{R}_0}$$

$$\text{tj. } \nabla_{R_0} = \frac{\partial \vec{R}}{\partial \vec{R}_0} \nabla_R + \sum_{i=1}^N \frac{\partial \vec{r}_i}{\partial \vec{R}_0} \nabla_{r_i} = \frac{M}{M + Nm} \nabla_R - \sum_{i=1}^N \nabla_{r_i} \quad \dots (1)$$

$$\frac{\partial \psi}{\partial \vec{R}_1} = \frac{\partial \psi}{\partial \vec{R}} \frac{\partial \vec{R}}{\partial \vec{R}_1} + \sum_i \frac{\partial \psi}{\partial \vec{r}_i} \frac{\partial \vec{r}_i}{\partial \vec{R}_1}$$

$$\nabla_{R_1} = \frac{m}{M + Nm} \nabla_R + \nabla_{r_1} \quad \text{tj. } \nabla_{R_i} = \frac{m}{M + Nm} \nabla_R + \nabla_{r_i} \quad \dots (2)$$

$$\nabla_{R_0}^2 = \left(\frac{M}{M + Nm} \right)^2 \nabla_R^2 - \frac{2M}{M + Nm} \sum_{i=1}^N \nabla_R \cdot \nabla_{r_i} + \left(\sum_i \nabla_{r_i} \right)^2$$

$$\nabla_{R_i}^2 = \left(\frac{m}{M + Nm} \right)^2 \nabla_R^2 + \frac{2m}{M + Nm} \nabla_R \cdot \nabla_{r_i} + \nabla_{r_i}^2$$

$$T = -\frac{\hbar^2}{2M} \left[\left(\frac{M}{M+Nm} \right)^2 \nabla_R^2 - \frac{2M}{M+Nm} \sum_i \nabla_R \cdot \nabla_{r_i} + \left(\sum \nabla_{r_i} \right)^2 \right] \\ + \sum_i \left\{ -\frac{\hbar^2}{2m} \left[\left(\frac{m}{M+Nm} \right)^2 \nabla_R^2 + \frac{2m}{M+Nm} \nabla_R \cdot \nabla_{r_i} + \nabla_{r_i}^2 \right] \right\}$$

$$= -\frac{\hbar^2}{2M} \frac{M+Nm}{(M+Nm)^2} \nabla_R^2 - \frac{\hbar^2}{2} \left[-\frac{2}{M+Nm} + \frac{2}{M+Nm} \right] \sum \nabla_R \cdot \nabla_{r_i} \\ - \frac{\hbar^2}{2M} \sum \nabla_{r_i}^2 - \frac{\hbar^2}{2M} 2 \sum_{i < j} \nabla_{r_i} \cdot \nabla_{r_j} - \frac{\hbar^2}{2m} \sum \nabla_{r_i}^2$$

$$T = -\frac{\hbar^2}{2(M+Nm)} \nabla_R^2 - \frac{\hbar^2}{2M} \sum_{i=1}^N \nabla_{r_i}^2 - \frac{\hbar^2}{M} \underbrace{\sum_{i < j} \nabla_{r_i} \cdot \nabla_{r_j}}_{\text{mass polarisation term}}$$

$$\mu = \frac{mM}{M+m}$$

$$V(\vec{r}_1, \vec{r}_2, \dots, \vec{r}_N) = -\sum_{i=1}^N \frac{Ze^2}{4\pi\epsilon_0 r_i} + \sum_{i < j} \frac{e^2}{4\pi\epsilon_0 r_{ij}}$$

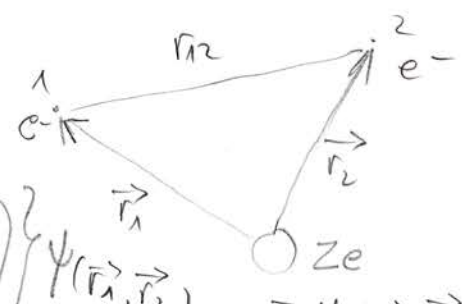
$$H \Psi_{\text{tot}}(\vec{R}, \vec{r}_1, \vec{r}_2, \dots, \vec{r}_N) = E_{\text{tot}} \Psi_{\text{tot}}(\vec{R}, \vec{r}_1, \vec{r}_2, \dots, \vec{r}_N)$$

$$\Psi_{\text{tot}}(R, r_1, \dots, r_N) = \Phi(R) \Psi(r_1, r_2, \dots, r_N)$$

$$-\frac{\hbar^2}{2(M+Nm)} \nabla_R^2 \Phi(R) = E_{\text{cm}} \Phi(R)$$

$$\left[-\frac{\hbar^2}{2M} \sum_{i=1}^N \nabla_{r_i}^2 - \frac{\hbar^2}{M} \sum_{i < j} \nabla_{r_i} \cdot \nabla_{r_j} + V(r_1, \dots, r_N) \right] \Psi = E \Psi(r_1, \dots, r_N)$$

$$E_{\text{tot}} = E_{\text{cm}} + E$$



$$\left\{ -\frac{\hbar^2}{2m} \nabla_1^2 - \frac{\hbar^2}{2m} \nabla_2^2 + \frac{e^2}{4\pi\epsilon_0} \left(-\frac{Z}{r_1} - \frac{Z}{r_2} + \frac{1}{r_{12}} \right) \right\} \psi(\vec{r}_1, \vec{r}_2) = E \psi(\vec{r}_1, \vec{r}_2)$$

Pod pretpostavom da je $M = \infty$, otako nije, tačnije izdat bi
 čuo.

$$\left[-\frac{\hbar^2}{2M} \nabla_{r_1}^2 - \frac{\hbar^2}{2M} \nabla_{r_2}^2 - \frac{\hbar^2}{M} \nabla_{r_1} \cdot \nabla_{r_2} - \frac{Ze^2}{4\pi\epsilon_0 r_1} - \frac{Ze^2}{4\pi\epsilon_0 r_2} + \frac{e^2}{4\pi\epsilon_0 r_{12}} \right] \psi(\vec{r}_1, \vec{r}_2) = E \psi(\vec{r}_1, \vec{r}_2)$$

$\mu = \frac{mM}{m+M}$; mass polarisation term

Zbog $\frac{1}{r_{12}}$ j-na nije separabilna i $\psi(\vec{r}_1, \vec{r}_2)$ se ne može
 napisati u vidu proizvoda jednoelektronskih t.f-ja.

T.f-je za sistem od dva ili više elektrona koje se
 ne mogu predstaviti kao single product of one-particle
 w.f. are said to be entangled. Odgovarajuća stanja
 su entangled: mereње ne može biti urađeno na jednoj
 čestici bez uticaja druge.

Primećimo da ψ -ne su nepravmejene ako se koordinate
 dva e^- promene. Neica je P_{12} an interchange operator

$$\psi(\vec{r}_2, \vec{r}_1) = P_{12} \psi(\vec{r}_1, \vec{r}_2)$$

Oko t.f-je zadovoljavaju istu šr. j-nu

$$\psi(\vec{r}_2, \vec{r}_1) = P_{12} \psi(\vec{r}_1, \vec{r}_2) = \lambda \psi(\vec{r}_1, \vec{r}_2) \quad \text{primenom opst } P_{12}$$

$$P_{12}^2 \psi(\vec{r}_1, \vec{r}_2) = \lambda P_{12} \psi(\vec{r}_1, \vec{r}_2) = \lambda^2 \psi(\vec{r}_1, \vec{r}_2) = \psi(\vec{r}_1, \vec{r}_2)$$

$$\Rightarrow \lambda^2 = 1 ; \lambda = \pm 1$$

$$\psi(\vec{r}_2, \vec{r}_1) = \pm \psi(\vec{r}_1, \vec{r}_2)$$

Simetričnost i antisimetričnost f-je
 izražava činjenicu da se čestice
 ne mogu razlikovati

+ → odgovara space-symmetric w.f. $\psi_+(\vec{r}_1, \vec{r}_2)$ or. para state
 - → -11- - -anti- $\psi_-(\vec{r}_1, \vec{r}_2)$ or. ortho state

Model nezavisnih čestica

He/2

Prost model koji daje mogućnost da kvalitativno računamo osnovne spektara (atomske jedinice)

$$H = H_0 + H'$$

$$H_0 = -\frac{1}{2} \nabla_{r_1}^2 - \frac{Z}{r_1} - \frac{1}{2} \nabla_{r_2}^2 - \frac{Z}{r_2} \equiv \hat{h}_1 + \hat{h}_2 \left[\hat{h}_i = -\frac{1}{2} \nabla_{r_i}^2 - \frac{Z}{r_i} \right] \\ (i=1,2)$$

a perturbacija 1^e $H' = \frac{1}{r_{12}}$

$$\hat{h}_i \psi_{n_i l_i m_i} = E_{n_i} \psi_{n_i l_i m_i} ; \quad E_{n_i} = -\frac{1}{2} \frac{Z^2}{n_i^2}$$

$$H_0 \psi^{(0)}(\vec{r}_1, \vec{r}_2) = E^{(0)} \psi^{(0)}(\vec{r}_1, \vec{r}_2)$$

$$\psi^{(0)}(\vec{r}_1, \vec{r}_2) = \psi_{n_1 l_1 m_1}(\vec{r}_1) \psi_{n_2 l_2 m_2}(\vec{r}_2)$$

$$E_{n_1 n_2}^{(0)} = E_{n_1} + E_{n_2} = -\frac{Z^2}{2} \left(\frac{1}{n_1^2} + \frac{1}{n_2^2} \right)$$

- Osnovno stanje -

$$\psi_0^{(0)}(\vec{r}_1, \vec{r}_2) = \psi_{1s}(r_1) \psi_{1s}(r_2) = \frac{Z^3}{\pi} \exp[-Z(r_1 + r_2)]$$

$$E_0^{(0)} = E_{n_1=1, n_2=1}^{(0)} = -Z^2 \text{ (a.u.)} = -4 \text{ a.u.} (\approx -108,8 \text{ eV})$$

Prva popravka energije osnovnog stanja:

$$E_0^{(1)} = \langle \psi_0^{(0)} | H' | \psi_0^{(0)} \rangle = \int |\psi_{1s}(r_1)|^2 \frac{1}{r_{12}} |\psi_{1s}(r_2)|^2 d\vec{r}_1 d\vec{r}_2$$

$$\text{ili u SI sistemu } E_0^{(1)} = \int |\psi_{1s}(r_1)|^2 \frac{e^2}{4\pi\epsilon_0 r_{12}} |\psi_{1s}(r_2)|^2 d\vec{r}_1 d\vec{r}_2$$

Interpretacija: $|\psi_{1s}(r_i)|^2$ gustina v-ice nabojeva e "1" u \vec{r}_i

$\rho(r_1) = -e |\psi_{1s}(r_1)|^2$ gustina nabojevanja zbog elektrona 1

$\rho(r_2) = -e |\psi_{1s}(r_2)|^2$

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=> poruci integral je elektrostatička interakcijska energija of two overlapping spherically symmetric distributions of electricity.

$$E_0^{(1)} = \frac{Z^6}{\pi^2} \int \exp[-2Z(r_1+r_2)] \frac{1}{r_{12}} d\vec{r}_1 d\vec{r}_2$$

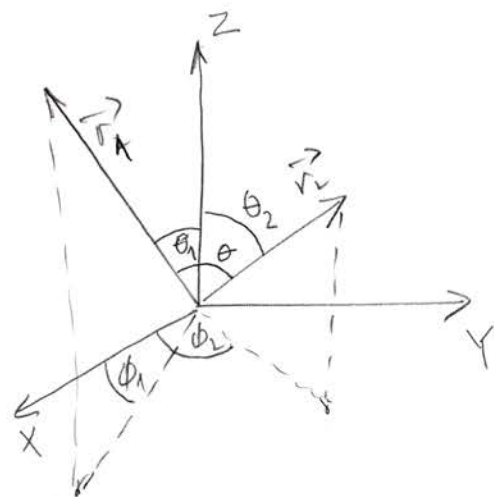
Za rešavanje koristimo generalnu proceduru koja je veoma korisna u atomskoj fizici.

Razvijmo $\frac{1}{r_{12}}$ u Legendrove polinome:

$$\begin{aligned} \frac{1}{r_{12}} &= \frac{1}{r_1} \sum_{l=0}^{\infty} \left(\frac{r_2}{r_1}\right)^l P_l(\cos\theta), \quad r_1 > r_2 \\ &= \frac{1}{r_2} \sum_{l=0}^{\infty} \left(\frac{r_1}{r_2}\right)^l P_l(\cos\theta), \quad r_1 < r_2. \end{aligned}$$

gde je θ ugao između \vec{r}_1 i \vec{r}_2 takođe je:

$$\cos\theta = \cos\theta_1 \cos\theta_2 + \sin\theta_1 \sin\theta_2 \cos(\phi_1 - \phi_2)$$



ili u kompaktnoj formi:

$$\frac{1}{r_{12}} = \sum \frac{(r_<)^e}{(r_>)^{e+1}} P_e(\cos\theta)$$

$r_<$ manje od r_1, r_2 a $r_>$ veće

Koristeći adicijsku teorem za sferne harmonike:

$$\frac{1}{r_{12}} = \sum_{l=0}^{\infty} \sum_{m=-l}^{+l} \frac{4\pi}{2l+1} \frac{(r_<)^e}{(r_>)^{e+1}} Y_{lm}^*(\theta_1, \phi_1) Y_{lm}(\theta_2, \phi_2)$$

jer je $P_l(\cos\theta) = \frac{4\pi}{2l+1} \sum_{m=-l}^{+l} Y_{lm}^*(\theta_1, \phi_1) Y_{lm}(\theta_2, \phi_2)$

Koristeći $Y_{00} = \frac{1}{\sqrt{4\pi}}$ imamo:

$$E_0^{(1)} = \frac{Z^6}{\pi^2} \sum_{l=0}^{\infty} \sum_{m=-l}^{+l} \frac{(4\pi)^2}{2l+1} \int_0^{\infty} dr_1 r_1^2 \int_0^{\infty} dr_2 r_2^2 \exp[-2Z(r_1+r_2)] \frac{(r_<)^e}{(r_>)^{e+1}} \times$$

$$\times \int d\Omega_1 Y_{lm}^*(\theta_1, \phi_1) Y_{00} \int d\Omega_2 Y_{00} Y_{lm}(\theta_2, \phi_2)$$

$$= \frac{Z^6}{\pi^2} \sum_{l=0}^{\infty} \sum_{m=-l}^{+l} \frac{(4\pi)^2}{2l+1} \int_0^{\infty} dr_1 r_1^2 \int_0^{\infty} dr_2 r_2^2 \exp[-2Z(r_1+r_2)] \frac{(r_<)^e}{(r_>)^{e+1}} \delta_{l,0} \delta_{m,0}$$

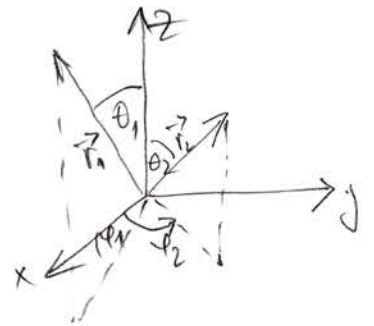
$\frac{1}{r_{12}}$ može da se razvije preko Legendre-ovih polinoma

$$\frac{1}{r_{12}} = \begin{cases} \frac{1}{r_1} \sum_{l=0}^{\infty} \left(\frac{r_2}{r_1}\right)^l P_l(\cos\theta), & r_1 > r_2 \\ \frac{1}{r_2} \sum_{l=0}^{\infty} \left(\frac{r_1}{r_2}\right)^l P_l(\cos\theta), & r_1 < r_2 \end{cases}$$

θ ugao između \vec{r}_1 i \vec{r}_2 : $\cos\theta = \cos\theta_1 \cos\theta_2 + \sin\theta_1 \sin\theta_2 \cos(\varphi_1 - \varphi_2)$

Ovo može da se zapiše i ovako

$$\frac{1}{|\vec{r}_1 - \vec{r}_2|} = \frac{1}{r_2} = \sum_{l=0}^{\infty} \frac{(r_2)^l}{(r_2)^{l+1}} P_l(\cos\theta)$$



Koristeći 'addition theorem'

$$P_l(\cos\theta) = \frac{4\pi}{2l+1} \sum_{m=-l}^{+l} Y_{lm}^*(\theta_1, \varphi_1) Y_{lm}(\theta_2, \varphi_2)$$

$$\frac{1}{r_2} = \sum_{l=0}^{\infty} \sum_{m=-l}^{+l} \frac{4\pi}{2l+1} \frac{(r_2)^l}{(r_2)^{l+1}} Y_{lm}^*(\theta_1, \varphi_1) Y_{lm}(\theta_2, \varphi_2)$$

Može da se pokaže i ovo [BJ. p. 614]

$$\frac{e^{ik|\vec{r}_1 - \vec{r}_2|}}{|\vec{r}_1 - \vec{r}_2|} = ik \sum_{l=0}^{\infty} (2l+1) j_l(kr_2) [j_l(kr_1) + i n_l(kr_1)] P_l(\cos\theta)$$

je $i n_l$ sfera Besselova i Neumann-ova f-ija.

i na kraju treba zneti i ovaj razvoj

$$e^{ik\vec{r}_1 \cdot \vec{r}_2} = \sum_{l=0}^{\infty} (2l+1) i^l j_l(kr) P_l(\cos\theta)$$

Sri salirei dvostrukie sume bide nula izuten $l=m=0$

$$E_0^{(1)} = 16Z^6 \int_0^\infty dr_1 r_1^2 \int_0^\infty dr_2 r_2^2 \exp[-2Z(r_1+r_2)] \frac{1}{r_1 r_2}$$

$$= 16Z^6 \int_0^\infty dr_1 r_1^2 \exp(-2Zr_1) \left[\frac{1}{r_1} \int_0^{r_1} dr_2 r_2^2 \exp(-2Zr_2) + \int_{r_1}^\infty dr_2 r_2 \exp(-2Zr_2) \right]$$

sada se lako radi i rešenje je

$$E_0^{(1)} = \frac{5}{8} Z \text{ a.u.}$$



Patili smo na dva dela
 $\frac{1}{r_2} = \frac{1}{r_1}$, $r_1 > r_2$
 $\frac{1}{r_2} = \frac{1}{r_2}$, $r_2 > r_1$
 Zapravo dvostruka
 suma $\frac{1}{r_1 r_2}$ za $l=m=0$
 to postaje

ili u SI sistemu

$$E_0^{(1)} = \frac{5}{8} \left(\frac{e^2}{4\pi\epsilon_0} \right) \frac{Z}{a_0}$$

Kako je He ; $Z=2$

$$E_0^{(1)} \approx 34 \text{ eV} = E_{1s^2}^{Z=2}$$

	$-E_0^{(0)}$ [eV]	$-E \approx E_0^{(0)} + E^{(1)}$ [eV]	$-E^{EXP}$ [eV]	$\frac{ E^{Calc} - E^{Exp} }{E^{Exp}} \%$
He	108,76	74,78	78,98	5,3
Li ⁺	244,71	193,73	198,04	2,2
Be ⁺⁺	435,04	367,06	371,51	1,2
B ³⁺	679,75	594,78	599,43	0,8
C ⁴⁺	978,84	876,88	881,83	0,6

Vidimo da teorija, koristeći samo prvu aproksimaciju, daje dosta dobro slaganje računskih vrednosti sa eksperimentalnim. Očigledno je slaganje bolje ukoliko je redni broj elementa viši.

Napomena:

Drugi način računanja integrala može ovako:
uzmemo z-osu duž vektora položaja \vec{r}_1 pa je onda:

$$r_{12} = \sqrt{r_1^2 + r_2^2 - 2r_1r_2 \cos\theta}$$

onda integracija po θ , može da se uvede pomoću:

$$\int \frac{\sin\theta d\theta}{\sqrt{r_1^2 + r_2^2 - 2r_1r_2 \cos\theta}} = \begin{cases} \frac{4}{r_2} & \text{za } r_1 < r_2 \\ 0 & , \quad r_1 > r_2 \end{cases}$$

Primeru teorije perturbacije uz popravke drugog reda nailazi na nepraktično glomazna izračunavanja. Dalja precizna izračunavanja dobijaju se drugim aproksimacijama prvenstveno vanjačevom metodom. (Sada vodi vanjačev metod)

$$\Psi(\vec{x}_1, \vec{x}_2) = \frac{\lambda^3}{\pi} e^{-\lambda(x_1+x_2)}, \quad \lambda = 1,6875 \quad \text{Hylleraas w.f.}$$

$$\Psi(x_1, x_2) = \frac{N}{\pi} (e^{-\alpha_1 x_1 - \alpha_2 x_2} + e^{-\alpha_2 x_1 - \alpha_1 x_2}) \left\{ \begin{array}{l} \text{Silverman et al. w.f.} \\ \text{[Chandrasekhar]} \end{array} \right.$$

$$N = \left[\frac{1}{\alpha_1^3} + \frac{1}{\alpha_2^3} + \frac{16}{(\alpha_1 + \alpha_2)^3} \right]^{-1/2}$$

Multi parameter (12-108), highly correlated w.f. of Joachain and Vanderpoorten a isti oblik imaju i f-je Byron and Joachain

$$\Psi(\vec{x}_1, \vec{x}_2) = \sum_{l=0}^{\infty} R_l(x_1, x_2) P_l(\hat{x}_1 \cdot \hat{x}_2)$$

$$R_l(x_1, x_2) = \frac{1}{4^{ll}} \sum_{nn'} A_{nn'}^{(l)} [x_1^{n+l} x_2^{n'+l} e^{-\frac{1}{2}(\alpha x_1 + \beta x_2)} + (1 \leftrightarrow 2)]$$

$A_{nn'}^{(l)}$, α, β → parametri;

F-je Tweed-a (od 21 do 41 parametra) su istog oblika samo

umesto $(x_1, x_2)^e$ / Druge NIMB (1996)
E = -2903724377 034119598 13 au.

Varijacijski metod

Probna
f- ψ :

$$\Psi(\vec{r}_1, \vec{r}_2) = \Psi_1(\vec{r}_1) \Psi_2(\vec{r}_2) = \frac{1}{\pi} \frac{Z e^3}{a^3} e^{-\frac{Z e r_1}{a}} e^{-\frac{Z e r_2}{a}}$$

$Z e$ - efektivna naelectrisacija tretiranu kao varijacijski parametar

Srednja vrednost energije

$$\bar{E} = \int \Psi^* \hat{H} \Psi d\tau$$

$$\hat{H} = -\frac{\hbar^2}{2m} (\nabla_1^2 + \nabla_2^2) - \frac{Z e^2}{4\pi\epsilon_0} \left(\frac{1}{r_1} + \frac{1}{r_2} \right) + \frac{e^2}{4\pi\epsilon_0 r_{12}}$$

$\Psi_1(\vec{r}_1)$ i $\Psi_2(\vec{r}_2)$ su rešenja za jednoelectronsice atome

$$-\frac{\hbar^2}{2m} \nabla_1^2 \Psi_1(\vec{r}_1) = \frac{1}{4\pi\epsilon_0} \frac{Z e e^2}{r_1} \Psi_1(\vec{r}_1) + Z e^2 E_H \Psi_1(\vec{r}_1)$$

$$-\frac{\hbar^2}{2m} \nabla_2^2 \Psi_2(\vec{r}_2) = \frac{1}{4\pi\epsilon_0} \frac{Z e e^2}{r_2} \Psi_2(\vec{r}_2) + Z e^2 E_H \Psi_2(\vec{r}_2)$$

$$Z e^2 E_H = -\frac{m Z e^2 e^4}{(4\pi\epsilon_0)^2 \hbar^2} = Z e^2 \underbrace{\left(-\frac{e^2}{(4\pi\epsilon_0) 2a} \right)}_{E_H}$$

$$\Rightarrow \hat{H} = \frac{1}{4\pi\epsilon_0} \frac{Z e e^2}{r_1} + Z e^2 E_H + \frac{1}{4\pi\epsilon_0} \frac{Z e e^2}{r_2} + Z e^2 E_H - \frac{Z e^2}{4\pi\epsilon_0} \left(\frac{1}{r_1} + \frac{1}{r_2} \right) + \frac{e^2}{4\pi\epsilon_0 r_{12}}$$

$$= 2 Z e^2 E_H + \frac{e^2}{4\pi\epsilon_0} (Z - Z e) \left(\frac{1}{r_1} + \frac{1}{r_2} \right) + \frac{e^2}{4\pi\epsilon_0 r_{12}}$$

$$\bar{E} = 2 Z e^2 E_H + \frac{e^2}{4\pi\epsilon_0} (Z - Z e) \underbrace{\int \Psi_1^*(\vec{r}_1) \Psi_2^*(\vec{r}_2) \left(\frac{1}{r_1} + \frac{1}{r_2} \right) \Psi_1(\vec{r}_1) \Psi_2(\vec{r}_2) d\tau_1 d\tau_2}_{I_1}$$

$$+ \frac{e^2}{4\pi\epsilon_0} \underbrace{\int \Psi_1^*(\vec{r}_1) \Psi_2^*(\vec{r}_2) \frac{1}{r_{12}} \Psi_1(\vec{r}_1) \Psi_2(\vec{r}_2) d\tau_1 d\tau_2}_{I_2}$$

$$I_1 = \int \psi_1^2(\vec{r}_1) \frac{1}{r_1} d\vec{r}_1 \underbrace{\int \psi_2^2(\vec{r}_2) d\vec{r}_2}_{=1} + \underbrace{\int \psi_1^2(\vec{r}_1) d\vec{r}_1}_{=1} \int \psi_2^2(\vec{r}_2) \frac{1}{r_2} d\vec{r}_2$$

$$= 2 \int \psi_1^2(\vec{r}_1) \frac{1}{r_1} d\vec{r}_1 = \frac{2Ze^3}{\pi a^3} \int_0^\infty \int_0^\pi \int_0^{2\pi} \frac{1}{r_1} e^{-\frac{2Ze r_1}{a}} r_1^2 dr_1 d\varphi_1 \sin\theta_1 d\theta_1$$

$$I_1 = \frac{2Ze}{a}$$

Drugi integral smo vec' izracunali

$$I_2 = \frac{5}{8} \frac{Ze}{a}$$

$$E_H = -\frac{e^2}{4\pi\epsilon_0} \frac{1}{2a}$$

$$\Rightarrow \bar{E} = 2Ze^2 E_H + \frac{e^2(Ze-Z)}{4\pi\epsilon_0} \frac{2Ze}{a} + \frac{e^2}{4\pi\epsilon_0} \frac{5}{8} \frac{Ze}{a} = \left[2Ze^2 - 4Ze(Ze-Z) - \frac{5}{4}Ze \right] E_H$$

$$\frac{\partial \bar{E}}{\partial Ze} = \left(4Ze - 8Ze + 4Z - \frac{5}{4} \right) E_H = 0$$

$$\Rightarrow \boxed{Ze = Z - \frac{5}{16}}$$

$$E_{\text{min}} = 2 \left(Z - \frac{5}{16} \right)^2 E_H = -\frac{1}{4\pi\epsilon_0} \frac{e^2}{a} \left(Z - \frac{5}{16} \right)^2$$

Razvijemo dobijeni izraz:

$$E = -\frac{1}{4\pi\epsilon_0} \frac{Z^2 e^2}{a} + \frac{1}{4\pi\epsilon_0} \frac{5}{8} \frac{Ze^2}{a} - \frac{e^2}{4\pi\epsilon_0 a} \left(\frac{5}{16} \right)^2$$

ovo daje teorija
perturbacija

$$\Delta E = 2 \left(\frac{5}{16} \right)^2 E_H \quad \text{ne zavisi od } Z$$

Razlika
(efikasnost)

The integrals are now straightforward, and yield the answer

$$E_0^{(1)} = \frac{5}{8}Z \quad \text{a.u.} \quad [6.65]$$

which is a positive contribution to the energy, as expected. We also remark that $E_0^{(1)}$ is linear in Z [2], while the unperturbed energy $E_0^{(0)} = -Z^2$ a.u. is quadratic in Z , so that the ratio $|E_0^{(1)}/E_0^{(0)}|$ decreases like Z^{-1} when Z increases. This is in line with our comments in the previous section concerning the decrease in relative importance of the electron-electron repulsion term $1/r_{12}$ with increasing Z .

Adding the first-order correction [6.65] to our zero-order result $E_0^{(0)} = -Z^2$ a.u., we find for the ground state energy E_0 the approximate value

$$E_0 \approx E_0^{(0)} + E_0^{(1)} = -Z^2 + \frac{5}{8}Z \quad \text{a.u.} \quad [6.66]$$

Both this 'first-order' result and the unperturbed energy $E_0^{(0)}$ are given in Table 6.3 for various two-electron atoms (ions), from the negative hydrogen ion H^- ($Z = 1$) to four times ionised carbon C^{4+} ($Z = 6$). The 'exact' [3] values E_0^{ex} are also tabulated. Except for H^- , the simple first-order perturbation approach yields quite good results. If we define ΔE , where

$$\Delta E = E_0^{\text{ex}} - (E_0^{(0)} + E_0^{(1)}) \quad [6.67]$$

Table 6.3 Values of the ground state energy E_0 of the Hamiltonian [6.2], for various two-electron atoms and ions (in atomic units)

	Ground state energy			'Exact'
	Unperturbed $E_0^{(0)}$ (equation [6.37])	First order $E_0^{(0)} + E_0^{(1)}$ (equation [6.66])	Simple variational $Z_e = Z - \frac{5}{16}$ (equation [6.79])	
H^-	-1	-0.375	-0.473	-0.528
He	-4	-2.750	-2.848	-2.904
Li^+	-9	-7.125	-7.222	-7.280
Be^{2+}	-16	-13.50	-13.60	-13.66
B^{3+}	-25	-21.88	-21.97	-22.03
C^{4+}	-36	-32.25	-32.35	-32.41

[2] The fact that $E_0^{(1)}$ is linear in Z may readily be understood by noting that each charge distribution in [6.56] contains a total charge $-e$ and extends over a region of space of linear dimension given approximately by $a = a_0/Z$. Their mutual interaction energy is therefore roughly given by $e^2/(4\pi\epsilon_0)a = Ze^2/(4\pi\epsilon_0)a_0$, which is indeed proportional to Z .

[3] The 'exact' results quoted in Table 6.3 are accurate values of the ground state energy E_0 of the Hamiltonian [6.2], obtained by using the Rayleigh-Ritz variational method with elaborate trial functions. Only a few of the presently available significant figures are given. Since these 'exact' values of E_0 must still be corrected for the motion of the nucleus, as well as relativistic and radiative corrections, they should not be confused with the experimental ground state energies E_0^{exp} .

as the difference between the 'exact' value E_0^{ex} and the first-order result [6.66], we see that the ratio $|\Delta E/E_0^{\text{ex}}|$ varies from about 5 per cent for He to 0.4 per cent for C^{4+} . It is interesting to note that this first-order perturbation treatment, carried out by Unsöld in 1927, was the first quantum mechanical calculation of the helium ground state. In view of the large discrepancies shown by the old quantum theory calculations, the relatively small difference of about 0.15 a.u. (≈ 4 eV) between his result and the experimental value was very promising.

The calculation of second and higher order corrections, $E_0^{(n)}$ ($n \geq 2$), is a much more difficult problem, to which we shall return further, after having discussed the Rayleigh-Ritz variational approach.

Variational method

We have seen in Section 2.8 that if H denotes the Hamiltonian of a quantum system, and ϕ a physically admissible trial function, the functional

$$E[\phi] = \frac{\langle \phi | H | \phi \rangle}{\langle \phi | \phi \rangle} \quad [6.68]$$

provides a variational principle for the discrete eigenvalues of the Hamiltonian. Moreover, it also yields a minimum principle for the ground state energy. That is,

$$E_0 \leq E[\phi] \quad [6.69]$$

For the case of two-electron atoms the Hamiltonian H (neglecting the motion of the nucleus and all but the Coulomb interactions) is given by [6.2]. Following the Rayleigh-Ritz variational method, we shall use trial functions ϕ depending on variational parameters, and carry out the variation $\delta E = 0$ with respect to these parameters.

It is apparent from the discussion of Section 6.4 that a basic defect of the 'zero-order' ground state wave function [6.35] (from which the first-order energy result [6.66] was obtained) is that each electron moves in the fully unscreened field of the nucleus. In order to take into account approximately the screening effect of each electron on the other one, we shall therefore choose a trial function of the form [6.47]. That is, (in a.u.)

$$\phi(r_1, r_2) = \frac{Z_e^3}{\pi} e^{-Z_e(r_1+r_2)} \quad [6.70]$$

or

$$\phi(r_1, r_2) = \psi_{1s}^{Z_e}(r_1)\psi_{1s}^{Z_e}(r_2) \quad [6.71a]$$

where

$$\psi_{1s}^{Z_e}(r) = \left(\frac{Z_e^3}{\pi} \right)^{1/2} e^{-Z_e r} \quad [6.71b]$$

and the 'effective charge' Z_e is considered as a variational parameter.

Pobudena stanja

Za početak razmotrimo jednostavno pobudena stanja simetričnu prostorna f-ja nultog reda l=0

$$\Psi_{\pm}^{(0)}(\vec{r}_1, \vec{r}_2) = \frac{1}{\sqrt{2}} [\Psi_{100}(\vec{r}_1) \Psi_{n\ell m}(\vec{r}_2) \pm \Psi_{n\ell m}(\vec{r}_1) \Psi_{100}(\vec{r}_2)]$$

$$E_{1,m}^{(0)} = -\frac{Z^2}{2} \left(1 + \frac{1}{n^2}\right) \text{ a.u. } \quad n \geq 2 \quad \left| \begin{array}{l} + \text{ para} \\ - \text{ orto} \end{array} \right.$$

Prva popravka [za diferencijal vodi B.J. p. 339]

$$E_{\pm}^{(1)} = \langle \Psi_{\pm}^{(0)} | H^1 | \Psi_{\pm}^{(0)} \rangle \quad ; \quad H^1 = \frac{1}{r_{12}}$$

$$E_{+}^{(1)} = J + K \quad ; \quad E_{-}^{(1)} = J - K$$

gde je $J = \int |\Psi_{100}(\vec{r}_1)|^2 \frac{1}{r_{12}} |\Psi_{n\ell m}(\vec{r}_2)|^2 d\vec{r}_1 d\vec{r}_2$ [Coulomb-ov ili direktni integral]

$$K = \int \Psi_{100}^*(\vec{r}_1) \Psi_{n\ell m}^*(\vec{r}_2) \frac{1}{r_{12}} \Psi_{100}(\vec{r}_2) \Psi_{n\ell m}(\vec{r}_1) d\vec{r}_1 d\vec{r}_2$$
 [exchange integral]

Kako je $\Psi_{n\ell m}(\vec{r}) = R_{n\ell}(r) Y_{\ell m}(\theta, \varphi)$ i koristeći ortogonalnost sf. harmonika i ratvoj za $\frac{1}{r_{12}}$ imamo:

$$J_{ne} = \int_0^{\infty} dr_2 r_2^2 R_{n\ell}^2(r_2) \int_0^{\infty} dr_1 r_1^2 R_{10}^2(r_1) \frac{1}{r_{>}}$$

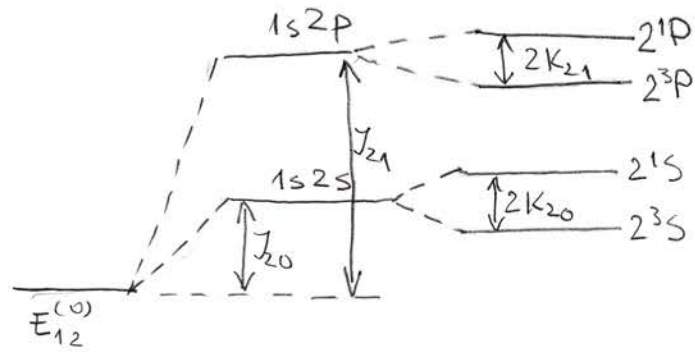
$$K_{ne} = \frac{1}{2\ell+1} \int_0^{\infty} dr_2 r_2^2 R_{10}(r_2) R_{n\ell}(r_2) \int_0^{\infty} dr_1 r_1^2 R_{10}(r_1) R_{n\ell}(r_1) \frac{(r_2)^{\ell}}{(r_{>})^{\ell+1}}$$

gde smo naglasili da svi integrali zavise od n i l a ne i od n

$$E_{ne\pm}^{(1)} = J_{ne} \pm K_{ne}$$

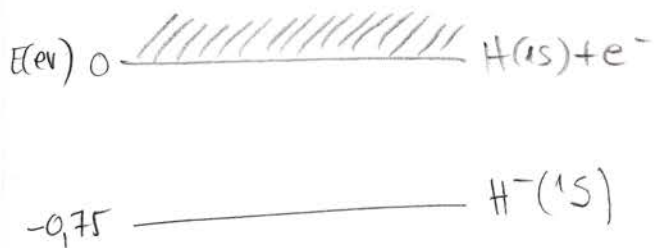
$$E_{ne\pm} = E_{1,m}^{(0)} + E_{ne\pm}^{(1)} \sim -\frac{Z^2}{2} \left(1 + \frac{1}{n^2}\right) + J_{ne} \pm K_{ne} \quad (*)$$

Integrali J_{ne} i K_{ne} mogu da se izračunaju eksplicitno, ovde ćemo navesti samo važne osobine. Pre svega J_{ne} mora biti pozitivno. Međutim, $K_{ne} > 0$ tako da sledi iz (*) orto stanje (tripletno) ima nižu energiju od odgovarajućeg para (splet) stanje što ima iste vrednosti n i l.



Radijativni prelazi između singletnog i tripletnog spinskog stanja (poznat kao interkombinacione linije) je zabranjen u dipolnoj aproksimaciji, što je spin-orbitovna interakcija se može zanemariti. To je slučaj sa atomima i jonsima sa niskim Z , tako da energ. spektar dvoelektr. atoma (jona) sa $Z \leq 40$ sadži dva približno nezavisna sistema nivoa: para i orto. (vidi sliku)

Spektar dvoelektr. jona sa $Z \gg 2$ je veoma sličan He-atomu. Sa druge strane negativni jon H^- ima samo jedno vezano stanje. Ionizac. potencijal je $0,75 \text{ eV}$ u odnosu na detachment u neutralni H i slobodan e^- .



Dvostruko pobuđena stanja

To su stanja kada su oba e^- pobuđena npr. $2s^2, 2s2p, 3p4d...$
Napomenimo da su ova stanja $\bar{l}\bar{l}$ iznad jonizacione
praga i zbog toga diskretna stanja ^(ubaceni) embedded
in the continuum. Tu se dešava radiationless transitions
iz dvostruko pobuđ. stanja na stanje jonizovane konfiguracije.
Takav prelaz je Auger-ov ili autojonizacioni, i
dvostruko pobuđ. stanja (koja su nestabilna u odnosu
na jonizaciju) se nazivaju autojonizaciona stanja
(B.J. p. 315]

Spinška stanja sa dva e^- mogu biti 4 f-je stanja

$$\begin{matrix} \alpha(1) \alpha(2) \\ \beta(1) \beta(2) \end{matrix}$$

$$\begin{matrix} \alpha(1) \beta(2) \\ \beta(1) \alpha(2) \end{matrix}$$

Osim te 4 moguće su i linearne kombinacije ovih 4 f-ja. Stanja $\alpha(1)\alpha(2)$ i $\beta(1)\beta(2)$ su simetrična a od stanja $\alpha(1)\beta(2)$ i $\beta(1)\alpha(2)$ možemo napraviti stanja koja će uvek biti simetrična ili antisimetrična

$$\alpha(1)\beta(2) + \alpha(2)\beta(1) \rightarrow \text{simetrična}$$

$$\alpha(1)\beta(2) - \alpha(2)\beta(1) \rightarrow \text{antisimetrična}$$

Ukupni m_s koji opisuje ove dve poslednje f-je je \emptyset a za stanja $\alpha(1)\alpha(2)$ je $+1$, dok za $\beta(1)\beta(2)$ -1

F-ja stanja spinka	$M_s = m_s(1) + m_s(2)$	S
$\alpha(1) \alpha(2)$	+1	1
$\alpha(1)\beta(2) + \alpha(2)\beta(1)$	0	
$\beta(1)\beta(2)$	-1	
$\alpha(1)\beta(2) - \alpha(2)\beta(1)$	0	0

$$\alpha(1) \alpha(2) \rightarrow m_s(1) + m_s(2) = \frac{1}{2} + \frac{1}{2} = 1$$

$$\beta(1) \beta(2) \rightarrow m_s(1) + m_s(2) = -\frac{1}{2} - \frac{1}{2} = -1$$

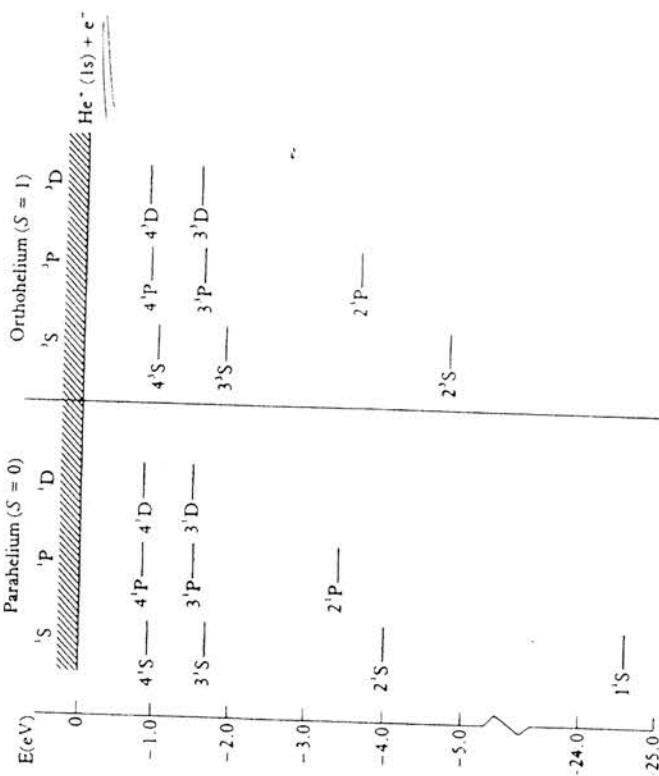
$$\alpha(1) \beta(2) \rightarrow m_s(1) + m_s(2) = +\frac{1}{2} - \frac{1}{2} = \emptyset$$

$$\beta(1) \alpha(2) \rightarrow m_s(1) + m_s(2) = -\frac{1}{2} + \frac{1}{2} = \emptyset$$

Spectroscopske oznake stanja $^2S^1L$ tako da
imamo singletno stanje $S=0$ i tripletno ($S=1$)

two-electron atoms

6.3



The experimental values of the lowest energy levels of helium. The energy scale is chosen so that 0 corresponds to the ionisation threshold. The configuration of each level is of the form $nl, (l+1)s + e^-$ doubly excited states (for example $2s, 2l$) are at positive energies on this scale, within the $(l+1)s + e^-$ continuum.

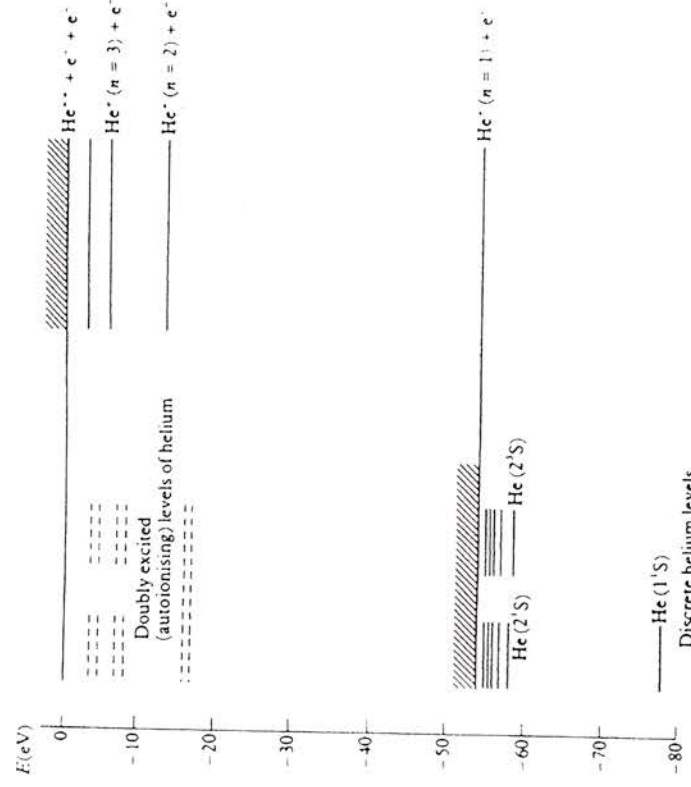
so on. In addition, a superscript to the left gives the value of the quantity $+1$, or multiplicity, which is equal to 1 for singlet ($S = 0$) states and 3 for l ($S = 1$) states.

We remark that Fig. 6.2 does not exhibit the fine structure splitting of the l s, due to the relativistic interaction between the spin and orbital angular momentum (spin-orbit effect) and to the magnetic interaction between the l s of the two electrons (spin-spin effect). Calling $J = L + S$ the total electric angular momentum and denoting by J^2 ($J + 1$) and M_J the eigenvalues of operators J^2 and J_z , respectively, it may be shown that the spin-orbit spin-spin interactions partially remove the degeneracy of the triplet states splitting them (except the 3S states) into three closely spaced levels corresponding to the three possible values $J = L + 1, L$ or $L - 1$ of the total angular momentum quantum number J . We shall discuss this problem in Chapter 8.

should be noted that Fig. 6.2 only represents the discrete part of the helium spectrum. A schematic diagram of the 'full' spectrum for the three-body system consisting of the He^{++} nucleus and two electrons is shown in Fig. 6.3. Choosing the origin of the energy scale in such a way that all three particles are

Level scheme of two-electron atoms

6.3



6.3 The 'complete' energy level spectrum of helium. The energy scale is relative to the threshold for the ionisation of both electrons and the zero of energy is 54 eV (the ground state energy of He^+), above the zero energy of the scale of Fig. 6.2.

unbound above $E = 0$, we see that the discrete levels of helium (displayed in more detail in Fig. 6.2) lie between the ground state value $E_0(\text{He}) \approx -79.0$ eV and the value $E_0(\text{He}^+) \approx -54.4$ eV of the ground state energy of the He^+ ion. Thus the ionisation potential numbers given in Fig. 6.2 correspond to the energy differences between the level $E_0(\text{He}^+)$ and a given energy level of the helium atom. For example, the ionisation potential corresponding to the helium ground state is

$$I_P = E_0(\text{He}^+) - E_0(\text{He}) \approx 24.6 \text{ eV} \quad [6.22]$$

The spectrum of two-electron ions with $Z > 2$ is similar to that of helium which we have just discussed. On the other hand, the negative hydrogen ion H^- , for which $Z = 1$, constitutes an interesting special case. Indeed, as shown on Fig. 6.4, this ion has only one bound state. The corresponding ionisation potential is about 0.75 eV, so that the H^- ion is barely stable against dissociation into a neutral hydrogen atom and a free electron. We shall return

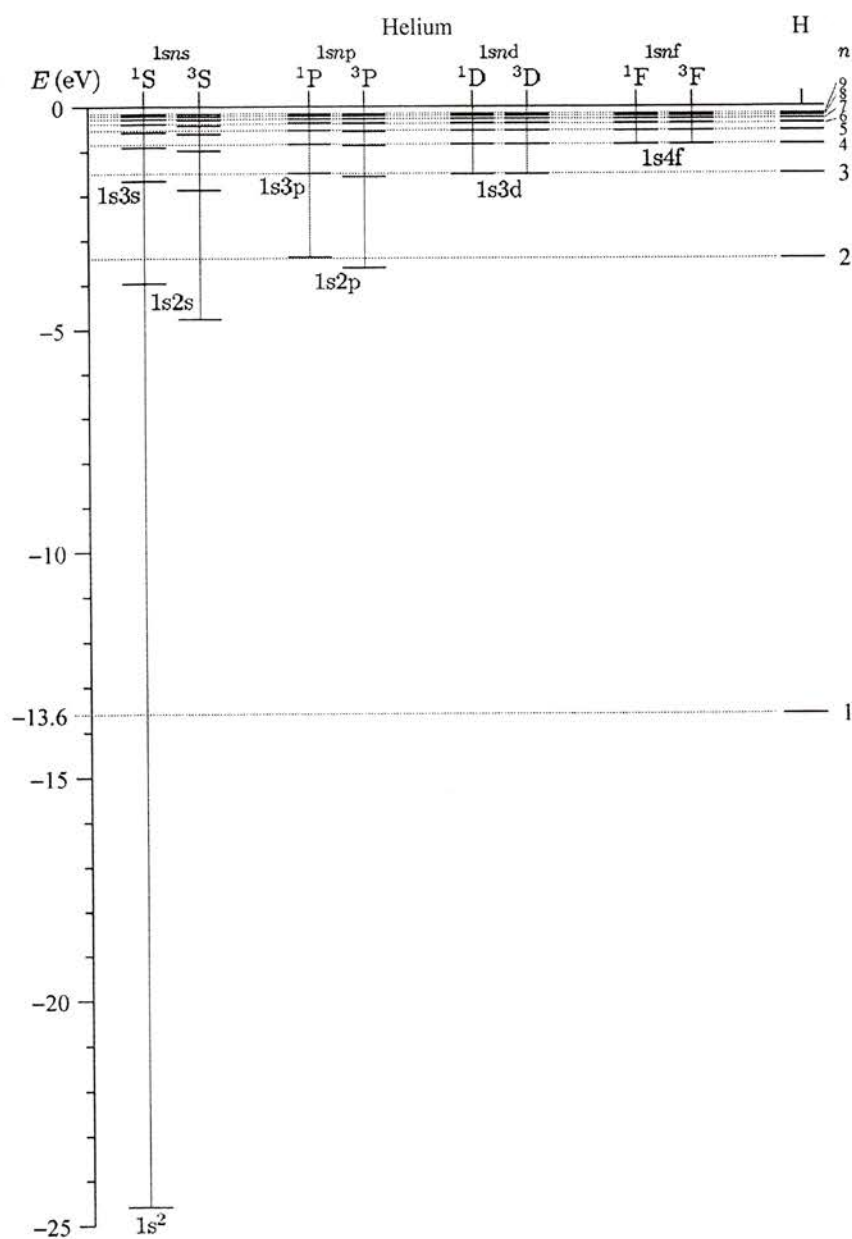


Fig. 3.4 The energy levels of the helium atom with those of hydrogen for comparison. The $1s^2$ ground configuration is tightly bound. For the excited configurations of helium the $1s$ -electron screens the outer electron from the nuclear charge so that the $1snl$ configurations in helium have similar energy to the shell with principal quantum number n in hydrogen. The hydrogenic levels are indicated on the right. The interval between the 1L and 3L terms (equal to twice the exchange integral) is clear for the $1s2s$, $1s2p$, $1s3s$, $1s3p$ and $1s4s$ configurations but it is smaller for higher n and l .